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PYRYLIUM SALTS FORMED BY DIACYLATION OF OLEFINS. XVIII. DIACETYLATION OF CYCLODODECENE

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Diacetylation of cyclododecene was accomplished by gradual addition of 70% perchloric acid (1 mole) into a stirred mixture of the olefin (1 mole) and excess acetic anhydride (5 moles) without cooling. Refluxing was caused by the exothermal reaction. After cooling, a double volume of diethyl ether was added, the upper layer was discarded, and the lower viscous black layer was extracted several times with boiling water. In the aqueous filtrates, straw-coloured needles of the diacetylation product of cyclododecene, 2,6-dimethyl-3,5-nonamethylene-pyrylium perchlorate (II), crystallised. They were purified by recrystallisation from 1% perchloric acid and had m.p. 170°. (Found: €,57.4; H,7.6; C<sub>16</sub>H<sub>24</sub>Cl O<sub>5</sub> requires €,57.9; H,7.3%).

III

II

I

The infrared spectrum in KBr-pellet presents CH stretching bands at 3027 (m), 2900 - 2950 (vs), and 2865 (s) cm<sup>-1</sup>, a band due to the  $\nu$ 8a vibration<sup>4</sup> at 1614 (s) cm<sup>-1</sup> and bands due to vibrations of the perchlorate anion and to deformation vibrations. In the NMR spectrum (tri-fluoroacetic acid solution) the six methyl protons appear as a singlet at  $\tau$  7.10, the four methylene protons (a) adjacent to the pyrylium ring as a triplet centered at  $\tau$  7.00 (coupling constant 6 cps), the next four methylene protons (b) as a symmetrical multiplet centered at  $\tau$  8.05, the other ten methylene protons (c) as a non-symmetrical multiplet at  $\tau$  ca. 8.75, and the pyrylium proton as a singlet at  $\tau$  1.26. These values are consistent with those obtained for other pyrylium salts<sup>5</sup> and thus serve as a reliable structural proof.

Treatment of the pyrylium salt (II) afforded a liquid pyridine (III) which was extracted with ether and distilled in vacuum. Its NMR spectrum presents in CCl<sub>4</sub> singlet peaks at  $\tau$  2.82 (one pyridine proton) and 7.62 (six methyl protons), a triplet (J=6.5 cps) centered at  $\tau$  7.32 (4 methylene protons of type a) and multiplets at  $\tau$  8.33 (4 methylene protons b) and 8.93 (ten methylene protons c).

Attempts to dibenzoylate cyclododecene with benzoyl chloride and aluminium chloride or stannic chloride were unsuccessful.

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